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Title: Advanced evaluation of landfill leachate treatments by low and high-resolution mass spectrometry focusing on microcontaminant removal

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Keywords: advanced oxidation processes; landfill leachate; mass spectrometry; organic microcontaminants; suspect and target analysis

Corresponding Author: Dr. Patricia Plaza-Bolaños, Ph.D.

Corresponding Author's Institution: University of Almeria

First Author: Ana Ruiz-Delgado, PhD student

Order of Authors: Ana Ruiz-Delgado, PhD student; Patricia Plaza-Bolaños, Ph.D.; Isabel Oller, PhD; Sixto Malato, PhD Research Professor; Ana Agüera, PhD Full Professor

Abstract: Conventional wastewater treatments are not usually effective in the remediation of specific landfill leachates due to their high content in toxic and recalcitrant compounds. Advanced and intensive treatments are needed for the decontamination and possible recycling of these effluents. Here, the combination of advanced oxidation processes (solar photo-Fenton) and an aerobic biological reactor have been applied to treat urban landfill leachates. The effectiveness of the proposed treatment line was also evaluated considering the removal of organic microcontaminants (OMCs) identified in the different phases, which is an innovative practice. The analytical strategy included: (i) a target approach (115 analytes) by liquid chromatography-mass spectrometry (LC- $\ensuremath{\mathsf{MS/MS}}\xspace$; and two suspect approaches using (ii) LC-high-resolution $\ensuremath{\mathsf{MS}}\xspace$ (database with >1300 compounds) and (iii) gas-chromatography-MS (database with >900 compounds). OMCs on the treated landfill leachate was reduced up to 94% of the initial concentration. 8 target compounds (mainly pharmaceuticals) out of 155 target analytes represented 85% of the OMC concentration in the raw leachate: cotinine, diclofenac, gabapentin, ketoprofen, lidocaine, mecoprop, nicotine and trigonelline. 3 nonpreviously reported OMCs were confirmed: di-n-nonyl phthalate, ophenylphenol and tonalide. Leachate partially oxidized by solar photo-Fenton process can be successfully incorporated to biological systems to complete the treatment by specifically adapted biomass.

1	Advanced evaluation of landfill leachate treatments by low and high-
2	resolution mass spectrometry focusing on microcontaminant removal
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5	A. Ruiz-Delgado ^{a,b} , P. Plaza-Bolaños ^{a,c,*} , I. Oller ^{a,b} , S. Malato ^{a,b} , A. Agüera ^{a,c}
6	
7	
8	^a CIESOL (Solar Energy Research Center), Joint Centre University of Almeria-CIEMAT,
9	Carretera de Sacramento s/n, E-04120, Almeria, Spain
10	
11	^b Plataforma Solar de Almería-CIEMAT, Carretera de Senés Km 4, E-04200 Tabernas,
12	Almeria, Spain
13	
14	^c Department of Chemistry and Physics, University of Almeria, Carretera de Sacramento
15	s/n, E-04120, Almeria, Spain
16	
17	*Corresponding author:
18	Email: pplaza.bolanos@ual.es; Tel: +0034 950 014139; Fax: +0034 950 015 008
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20 Abstract

Conventional wastewater treatments are not usually effective in the remediation of 21 specific landfill leachates due to their high content in toxic and recalcitrant compounds. 22 23 Advanced and intensive treatments are needed for the decontamination and possible 24 recycling of these effluents. In this studyHere, the combination of advanced oxidation processes (solar photo-Fenton) and an aerobic biological reactor have been applied to 25 treat urban landfill leachates. The effectiveness of the proposed treatment line was also 26 evaluated considering the removal of organic microcontaminants (OMCs) identified in 27 28 the different phases, which is an innovative practice. The analytical strategy included: (i) a target approach (115 analytes) by liquid chromatography-mass spectrometry (LC-29 30 MS/MS); and two suspect approaches using (ii) LC-high-resolution MS (database with >1300 compounds) and (iii) gas-chromatography-MS (database with >900 compounds). 31 32 OMCs on the treated landfill leachate was reduced up to 94% of the initial quantified 33 concentration. 8Eight target compounds (mainly pharmaceuticals) out of 155 target 34 analytes represented 85% of theis OMC concentration in the raw leachate: cotinine, diclofenac, gabapentin, ketoprofen, lidocaine, mecoprop, nicotine and trigonelline. 35 36 Three <u>3</u> non-previously reported OMCs were confirmed: di-n-nonyl phthalate, ophenylphenol and tonalide. The study demonstrated that lLeachate partially oxidized by 37 solar photo-Fenton process can be successfully incorporated to biological systems to 38 complete the treatment by means of specifically adapted biomass. 39

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Keywords: advanced oxidation processes; landfill leachate; mass spectrometry; organic
microcontaminants; suspect and target analysis.

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44 **1. Introduction**

Urban landfills are the destination of many products and household waste. The 45 leachates generated in controlled landfills are collected for their subsequent treatment 46 considering them as wastewater. As complex and hazardous wastewater, landfill 47 leachates contain high levels of toxic and recalcitrant compounds, including organic 48 microcontaminants (OMCs) considered highly threatening to the environment and 49 human health [1]. Consequently, landfill leachates are not efficiently treated by 50 conventional processes [1]. The release to the environment of leachates non-adequately 51 52 treated leads to the contamination of groundwater as a first consequence [2].

53 In general, leachates from landfills in methanogenic phase (stabilized and relatively 54 old) are specially bio-recalcitrant with a very low biological oxygen demand (BOD)/ chemical oxygen demand (COD) ratio (<0.1) and COD values below 3000 mg/L [3]. On 55 56 the contrary, young landfills produce leachates with considerably higher CODs (around 57 10000 mg/L) and with low to medium biodegradability [4]. Therefore, conventional 58 physical-chemical and biological treatments could be inefficient also for the remediation of young landfill leachates due to the possible presence of biorefractory compounds 59 [5,6]. In consequence, new advanced remediation technologies are required to face the 60 treatment of this type of complex wastewater. In this sense, in the last years, advanced 61 oxidation processes (AOPs) have been applied as a potential alternative strategy to 62 conventional technologies to treat leachates [7,8]. Among the different AOPs, the solar 63 photo-Fenton process (SPF) is currently one of the most attractive treatments for 64 different kinds of complex wastewaters, considering reaction rate, operating costs and 65 environmental impact [9]. More in detail, the possible combination of a SPF process 66 with a biological post-treatment, once improved the biocompatibility of leachates, will 67 68 mean a significant reduction on related operating costs with the added value of using a 69 renewable source of UV photons [10,11].

OMCs have been investigated in landfill leachates by applying different strategies. In 70 71 general, a target approach has been used and the focus has been put on a specific class of organic contaminants [12–18]. Other studies increased the scope to several families 72 73 with a variable number of monitored target compounds [7,19,20]. However, when using 74 target strategies, there is always missed information that can only be increased with the use of non-target and screening approaches. Some studies have applied liquid 75 76 chromatography coupled to low resolution mass spectrometry (LC-LRMS) and gas 77 chromatography coupled to LRMS (GC-LRMS) in the full scan mode to carry out

screening analysis of leachates [1,21]. Other authors performed the characterization of 78 landfill leachates using a screening strategy by GC coupled to high-resolution MS (GC-79 HRMS), focusing on GC-amenable OMCs [22]. However, sensitivity of HRMS 80 instruments cannot be compared to that of the modern LRMS instruments (triple 81 quadrupole or quadrupole-lineal ion trap, QqLIT). Bearing this in mind, the application 82 of combined target and non-target approaches using both HRMS and LRMS and LC 83 and GC seem to be an adequate strategy to deal with complex landfill leachates and 84 85 wide-scope analyses.

86 The present study aims the characterization of landfill leachates which were subjected to a combination of a physical-chemical pre-treatment, an AOP (SPF) and an aerobic 87 biological treatment. Therefore, the effectiveness of the decontamination line proposed 88 in each separate stage was evaluated not only by monitoring classical parameters (such 89 90 as COD, TOC, biodegradability, etc.), but also enriched with a detailed study including several advanced analytical techniques for OMC determination. Thus, raw leachate and 91 92 the effluents obtained after coagulation/flocculation step, SPF pre-oxidation and the final aerobic biological treatments were submitted to target analysis by LC-QqLIT-93 94 MS/MS, suspect analysis by LC coupled to quadrupole-time-of-flight MS (LC-QTOF-MS, for LC-amenable compounds) and screening analysis by GC coupled to quadrupole 95 MS (GC-Q-MS, for GC-amenable compounds). The information was used to evaluate 96 97 the performance of the whole proposed treatment line with respect to OMC elimination.

98

99 2. Experimental

100 2.1. Leachate samples origin and characterization

101 Landfill leachate samples were collected from an urban landfill placed on Vila Real (North of Portugal). This landfill receives 75000 ton/year of municipal solid wastes for 102 a population of around 50000 people and it started operating in the year 2000. The 103 104 leachates used in this study presented a pH of 8.5, 95.7 NTU of turbidity and 31.4 105 mS/cm of conductivity. The values of COD, total nitrogen (TN) and dissolved organic carbon (DOC) were 9387 mg O₂/L, 4854 mg N/L and 1860 mg C/L, respectively. These 106 leachates showed the following ionic concentrations: 2603 mg/L of Cl⁻, 3048 mg/L of 107 NH_4^+ , 84 mg/L of NO_3^- , 117 mg/L of PO_4^{3-} and 211 mg/L of SO_4^{2-} , apart from other 108 inorganic ions. 85% of acute toxicity (and non-biodegradable) was measured by 109

activated sludge from a municipal wastewater treatment plant (MWWTP) in arespirometer (by monitoring oxygen uptake rate) [23].

- 112
- 113 2.2. Analytical methods

114 2.2.1. Landfill leachate characterization

DOC was measured with a Shimadzu TOC-VCN analyzer (Kyoto, Japan) after 115 116 filtering samples through 0.2-µm syringe-driven filters. TN was quantified in a TC-TOC-TN analyzer coupled with a TNM-1 unit (model TOC-VCSH, Shimadzu). Anion 117 concentrations were determined with a Metrohm 872 Extension Module 1 and 2 ion 118 chromatography (IC) system (Herisau, Switzerland) configured for gradient analysis. 119 Cation and ammonium concentrations were determined with a Metrohm 850 120 Professional IC configured for isocratic analysis. COD was determined using 121 122 commercial kits (Spectroquant[™], Merk, Darmstadt, Germany). Total iron concentration was measured using the 1,10-phenantroline method following ISO 6332. Hydrogen 123 peroxide was analyzed by using titanium (IV) oxysulfate according to DIN 38402H15. 124 Turbidity was measured in a Hach 2100 N turbidimeter (Loveland, CO, USA). Sulfuric 125 acid (95-97%, J.T. Baker, Deventer, The Netherlands), hydrogen peroxide (35% w/v, 126 reagent-grade, J.T. Baker), titanium oxysulphate (1.9-2.1%, Sigma Aldrich, Steinheim, 127 Germany) and ferric chloride anhydrous (97%, Panreac, Barcelona, Spain) were also 128 129 used.

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131 2.2.2. Target analysis of OMCs by LC-QqLIT-MS/MS

132 Samples were analyzed according to a previously validated method using the direct injection (DI) technique [24]. This method was able to monitor 115 OMCs including 133 134 pharmaceuticals and pesticides at the ng/L level. Briefly, samples were filtered (0.22 μm, PTFE, Millipore, Cork, Ireland) and injected directly into the LC system. ¹³C-135 136 Caffeine was used as injection standard (purity >98%, Sigma-Aldrich). Appropriate dilution (normally 1:10 or 1:100 v/v) was applied when signal saturation was observed. 137 The chromatographic separation was accomplished using a Zorbax Eclipse XDB C18 138 analytical column (50 mm \times 4.6 mm, 1.8-µm particle size, Agilent Technologies, Foster 139 140 City, CA, USA). The LC-QqLIT-MS/MS system consisted of an Agilent 1200 141 chromatograph (Agilent Technologies) coupled to a 5500 QTRAP™ from Sciex (Foster City, CA, USA) equipped with an electrospray (ESI) source (TurboIon[™] Spray). The 142

Schedule MRM Algorithm[™] was applied for data acquisition (SRM mode, selected
reaction monitoring). More information regarding the LC gradient, MS/MS conditions
and software has been already reported [24].

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147 2.2.3. Suspect analysis of OMCs by LC-QTOF-MS

To increase the scope of the analysis, a subsequent determination using HRMS was 148 149 carried out. Filtrated leachate samples (0.22 µm, PTFE) were analyzed using a 150 previously developed strategy [25] and also a DI technique. Briefly, a suspect list of >1300 OMCs frequently found in wastewater was used to investigate the presence of 151 other OMCs out of the target method [26]. The separation by LC was performed by a 152 Poroshell 120 EC-C18 column (50 mm x 4.6 mm, 2.7-µm particle size, Agilent 153 Technologies). The LC-QTOF-MS system consisted of an Agilent 1260 chromatograph 154 155 (Agilent Technologies) coupled to a Triple TOF 5600+™ from Sciex equipped with a dual source (ESI for sample injection and atmospheric-pressure chemical ionization 156 (APCI) for calibrant delivery). The TOF-MS survey scan mode followed by four IDA 157 events (information dependent acquisition) were applied (m/z 100-2000; mass resolving 158 power, 30000). Further information about LC gradient, MS settings, screening 159 workflow and software can be found in the cited reference. 160

161

162 2.2.4. Suspect analysis of OMCs by GC-Q-MS

163 500 mL of raw samples were extracted by liquid-liquid extraction (LLE) with dichloromethane and *n*-hexane (see Supporting Material for extraction method details). 164 165 4 different extracts with a concentration factor of 500 were obtained for each leachate sample in order to cover a wide range of polarity and to extract as many OMCs as 166 167 possible. Samples were submitted to analysis in a GC-Q-MS system for the suspect analysis of >900 compounds. The GC-MS system consisted of a 7890B chromatograph 168 169 interfaced to a quadrupole (Q) analyzer 5977A (Agilent Technologies, Palo Alto, CA, 170 USA). The separation was carried out with an HP-5MS UI capillary column (30 m \times 171 0.25 mm i.d. \times 0.25 µm film thickness). Retention times (RT) were locked using chlorpyrifos methyl. Data was acquired in the full scan mode. Identification was carried 172 173 out using the data from the commercial library (Agilent Technologies) [27] and the 174 NIST library 2.2 (Ed. 2014). This database includes pesticides, endocrine disruptors, 175 transformation products, musk compounds, and persistent organic pollutants (POPs),

among others. It applies two main features: (i) the use of locked RTs so that the RTs of
the compounds included in the library can be precisely reproduced when using the same
column; and (ii) the application of mass spectral matching and deconvolution software
tools. Additional information can be found in the Supporting Material.

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181 *2.3. Treatment line for landfill leachate*

182 All studies were performed at pilot plant scale at Plataforma Solar de Almería (PSA, 183 Spain). It began with a coagulation/flocculation step by using ferric chloride anhydrous 184 at 0.5 g/L according to previous experience with similar wastewaters [11,28-30]. The 185 plant consisted of a 200-L conical mixer coupled to a filtration system: a silex filter of 75 μ m and two microfilters of 25 μ m and 5 μ m. Firstly, the adjustment of the raw 186 187 landfill leachate at pH 5 was carried out. Then, ferric chloride anhydrous was added and 3 min of rapid mixing at 100 rpm for coagulation, slow mixing at 30 rpm during 30 min 188 for flocculation, and 30 min of settling provoking solids sedimentation were applied. 189

Afterwards, SPF process (pH adjusted to 2.8-3 with sulfuric acid) was studied as a 190 191 second stage in a compound parabolic collector (CPC) pilot plant. This pilot plant (total 192 volume: 82 L) consists of a recirculation tank with one centrifugal pump for water flow, 4.16 m² of CPC total irradiated area, being 44.6 L the illuminated volume inside the 193 solar collector absorber tubes [9]. The addition of iron was not necessary as a 194 of 30 mg/L of Fe (III) 195 concentration remained dissolved after the coagulation/flocculation step. An initial dose of 200 mg/L of hydrogen peroxide was 196 197 added, mixed in the dark and the CPC uncovered, starting the SPF process. Hydrogen 198 peroxide was continuously added as consumed in the reaction bulk. The temperature inside the reactor was set and controlled at 35 °C. 199

200 The pH of the SPF effluent was adjusted to 7 and the treatment was completed in an 201 advanced biological pilot plant based on an aerobic immobilized biomass reactor (IBR) 202 operated in continuous mode. This system is composed of a 200-L conditioning tank and a 20-L recirculation tank connected to the 20-L IBR filled in with K1 supports 203 204 (AnoxKaldnes, diameter: 9.1 mm; surface area for biofilm growth: approximately $500 \text{ m}^2/\text{m}^3$; density 0.95 kg/dm³) that were colonized by active sludge from the 205 206 MWWTP of El Toyo (Almeria) and a final 20-L collecting conic tank. Biomass 207 adsorbed on the supports generated biofilms after 4 days of feeding with influent of the 208 MWWTP. Dissolved oxygen (always >2 mg/L), pH and temperature were recorded209 automatically.

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211 **3. Results and Discussion**

212 *3.1. Landfill leachate treatment*

Coagulation/flocculation pre-treatment, after the addition of 0.5 g/L of anhydrous 213 214 FeCl₃, reduced 53% of the turbidity and eliminated 30% of the initial DOC. The 215 subsequent SPF process was extended till a drastic reduction of toxicity and improvement of biodegradability were attained (data not shown) and thus, a 216 combination with an aerobic biological treatment could be considered feasible to 217 complete the treatment. At this stage, 43% of DOC was eliminated (final DOC of 218 219 750 mg/L) after 6.7 h of illumination time (Fig. 1). The partially treated landfill leachate 220 showed an increase in biodegradability (from 0.06 to 0.62) which was enough to start 221 the adaptation phase of an aerobic biological process and to finish the wastewater treatment. For that purpose, increasing volumes of the pre-treated landfill leachates 222 were fed (mixed with the influent of a conventional MWWTP) to the biological reactor 223 in batch mode. After 12 days of operation the complete adaptation of the biosystem was 224 225 attained and continues mode operation in the bioreactor began. Then, feeding flow was continuously increased from 0.16 L/h until attaining the maximum treatment capacity at 226 227 0.58 L/h after 10 days of operation (2.8 days of hydraulic retention time), obtaining a 228 maximum DOC elimination rate of 1.9 mg/L·min, and TN elimination rate of 8.3 mg/L· 229 min.

230 Different studies have reported also the combination of AOPs with aerobic biological 231 systems for landfill leachates treatment obtaining similar results to those shown in the 232 present work. Combination of technologies previously assessed by other authors for the treatment of landfill leachate were photo-Fenton [31] or electro-Fenton [32] with 233 biological treatments, Fenton combined with passive aerated inmobilized biomass 234 235 reactor [33], photo-electro-Fenton with membrane bioreactors [34], etc., all of them showing COD elimination percentages from 83 to 98% in line with the results observed 236 in the treatment line proposed in this work (97% of total COD removal). Specifically, 237 better results in terms of biodegradabilidy enhancement were obtained with the 238 239 application of photo-Fenton process in the proposed treatment line compared to $O_3/H_2O_2(0.29)$ [35]. 240

241

242 3.2. Target analysis of OMCs by LC-QqLIT-MS/MS

243 Leachate and effluents from the three treatment stages were submitted to analysis 244 (raw, coagulated-flocculated, SPF and IBR) using the method previously described to monitor 115 OMCs [24]. The main advantage of the DI technique is the absence of 245 246 losses in sample extraction. The DI technique is adequate when using sensitive MS 247 systems, such as the QqLIT analyzer, and considering that for many of these samples, a 248 dilution stage was required. The results are summarized in Fig. 2. Different types of 249 OMCs (up to 43 compounds, 37% of the monitored OMCs) were detected in the 250 samples (Table A.1, Supporting Information) at concentrations from ng/L to µg/L. A significant decrease in the OMC load was observed during the treatment stages, with a 251 252 total removal of 94%, from an initial total load in the raw leachate of 624.3 µg/L.

8 out of the 43 OMCs in the raw leachate represent 85% of the total load: cotinine 253 254 (nicotine metabolite), diclofenac (nonsteroidal anti-inflammatory agent), gabapentin 255 (antiepileptic drug, treatment of neuropathic pain), ketoprofen (nonsteroidal anti-256 inflammatory agent), lidocaine (local anesthetic, anti-arrhythmia agent), mecoprop 257 (herbicide), nicotine (toxic alkaloid from tobacco) and trigonelline (metabolite of niacin 258 or vitamin B3, biomarker for coffee consumption, antidiabetic properties, also found in 259 legumes and soy products). Some of these compounds (cotinine, lidocaine and nicotine) 260 were also reported at very high levels (>50 μ g/L) and high frequency of detection in 261 leachates from the U.S. [20] The 8 OMCs pointed out also represented 87% of the total 262 load in the coagulated-flocculated leachate. After the SPF stage (87% of removal), three 263 out of these eight compounds represented 98% of the total (gabapentin, nicotine and trigonelline). After the biotreatment (94% removal) still gabapentin, nicotine and 264 265 trigonelline represented 60% of the load but a substantial elimination was obtained in the treatment. The concentration of gabapentin, nicotine and trigonelline was 319.5, 266 267 27.9 and 37.2 µg/L, respectively, in the raw leachate and 15.1, 4.4 and 1.1 µg/L, after the biotreatment. 268

It is important to notice that the SPF effluent was transferred to an aerobic IBR, being mixed during continuous operation with influent from a MWWTP for a proper adaptation of the biomass. Consequently, the leachate was diluted, firstly with the aim of adapting the biomass and secondly, to enhance the treatment efficiency. The mixture with the influent of the MWWTP provoked a small increment of the concentrations of some OMCs in the outlet of the IBR (e.g. gabapentin, Table A.1). In any case, the
degradation of OMCs from leachate throughout the different treatment stages was still
higher than 94%. Indeed, from the 34 OMCs detected after the biotreatment stage (see
Table A.1), 27 came from influent of the MWWTP mixed with effluent from SPF (note
effluent from SPF contained only 7 OMCs), but a very low concentration levels.

279 In summary, the OMC content in the raw landfill leachate was clearly eliminated after the treatment line, with a final global concentration of $34.440.5 \,\mu$ g/L, similar to the one 280 that can be measured in a typical urban biological effluent (primary + secondary 281 282 treatment).[24,316] Despite only a few works have been published about this topic, the proposed treatment line achieved better effectiveness, in terms of OMCs degradation, 283 284 than the results reported by, for instance, Rocha et al. [7]. In this study, the presence of 285 40 OMCs (considering GC and LC-amenable analytes) were confirmed in leachates 286 submitted to a photo-Fenton process, with a complete elimination of only 24 OMCs.

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288 3.3. Suspect analysis of OMCs by LC-QTOF-MS

289 In order to increase the scope of the OMC identification study, the suspect screening 290 previously explained was applied. The list of >1300 chemicals were compiled in the 291 NORMAN suspect list exchange [26] and reported in literature. After assessing the 292 compliance with the proposed identification criteria [25], analytical standards for 4 suspect compounds were purchased for a conclusive confirmation (Table 1). The 293 294 selected candidates were only found in the coagulated-flocculated and biological 295 effluents (due to the already commented mix during continuous operation with typical 296 influent from MWWTP). Some of them must be also present in the raw leachate, but 297 they were not probably detected because of an intense matrix effect. No suspect 298 compounds were found in the SPF effluent. After the injection of the reference 299 standards, the RT could be evaluated and the following compounds were confirmed in 300 the coagulated-flocculated effluent (Fig. A.1): chlorfenvinphos (pesticide), niflumic 301 acid (analgesic and anti-inflammatory pharmaceutical), tramadol (opioid analgesic) and 302 valsartan (treatment of hypertension).

The variation of the levels of the confirmed compounds after each stage of the treatment line is shown in Table A.2. Removal was >99% for chlorfenvinphos and niflumic acid, whereas lower removals were obtained for tramadol (60%) and valsartan (21%). The full scan chromatogram obtained by LC-QTOF-MS indicated a general decrease of the number of features detected by the system because of the elimination ofthe organic content, which includes OMCs (Fig. 3).

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310 3.4. Suspect analysis of OMCs by GC-Q-MS

Leachate samples had to be extracted and transferred to an organic solvent prior to GC 311 312 analysis. A typical LLE scheme (Fig. A.2) was applied since it is a non-specific 313 extraction technique and it allows a high pre-concentration of the sample to increase 314 sensitivity. In this case, the Q system operating in the full scan mode is not as sensitive 315 as the QqLIT analyzer operating in the SRM mode, and a preconcentration stage is 316 mandatory. In this sense, LLE is commonly used in different EPA methods for OMC 317 determination, such as EPA Method 508A [372] (screening of polychlorinated 318 biphenyls), EPA Method 508 [338] (analysis of chlorinated pesticides) or the EPA Method 1613 [349] (dioxin analysis) and it is usually considered as a reference 319 extraction technique providing a high sample preconcentration. In order to cover a wide 320 321 range of polarities, the combination of two different extraction solvents, such as 322 dichloromethane and *n*-hexane, was applied [22]. Besides, a sequential extraction was 323 also used to increase the extraction capability and matrix removal to reduce matrix 324 effect (Fig. A.2). Together with the samples, a reagent blank (analysis performed using 325 reagents and solvents, without sample, which is substituted by MilliQ water) was also extracted to evaluate the background levels of certain substances which may be present 326 327 in the leachate samples, such as phthalates.

328 The selection of the suspect candidates (Table 2) was accomplished based on the following criteria: (i) the compound was searched in a RT window of \pm 0.350 min, 329 except for certain compounds typically showing a high peak width or groups of 330 331 compounds such as pesticide triadimenol (±0.700 min); (ii) a tolerance of 40% was applied on the relative responses of the qualifiers ions; (iii) a minimum MS library 332 333 match score of 60% was required to evaluate the compound as a possible candidate. The occurrence of the same suspect compound in the different extracts of the same sample 334 was also considered, as well as the peak shape, which can be very characteristic for 335 certain compounds (e.g. naphthalene). In general, the specific extractions with n-336 337 hexane, and the sequential extractions with dichloromethane were able to show a higher number of peaks. The list of 19 suspect candidates included plasticizers, pesticides, 338 PAHs and musk compounds, among others (Table 2). Commercial standards of the 339

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340 selected candidates were acquired and injected in the GC-MS system. 12 out of the 19 candidates were finally confirmed in the samples with concentrations in the range from 341 342 0.02 (phenanthrene, PAH) to 4.9 µg/L (diisobutyl phthalate, plasticizer) as it is shown in 343 Fig. 4. Some compounds showed higher concentrations in the coagulated-flocculated 344 effluent than in the raw samples, probably due to a higher matrix effect that induce a signal suppression effect. In any case, a decrease of the OMC concentration at the end 345 of the treatment line was obtained. As expected, phthalate compounds showed very high 346 concentrations, due to their extended use as plasticizers. To our knowledge, 3 347 348 compounds have not been previously reported: di-n-nonyl phthalate (plasticizer), ophenylphenol (pesticide) (Fig. A.3), and tonalide (musk compound). 349

350 When comparing the results with previous studies, the occurrence of phthalate 351 compounds has been often reported [1]. Four phthalate compounds were detected in the 352 raw leachate, ranging from 0.2 (butyl benzyl phthalate) to 4.9 µg/L (diisobutyl 353 phthalate). These concentrations were comparable to the results reported by Castillo et 354 al. [21] (di-*n*-butylphthalate, $<1 \mu g/L$) and Marttinen et al. [18] (butyl-benzyl phthalate, $<1 \mu g/L$ and di-*n*-butylphthalate, $<20 \mu g/L$); Rocha et al. [7] reported considerably 355 356 higher levels (<500 µg/L) for these two phthalates. In the case of PAHs, naphthalene 357 and phenanthrene were confirmed at levels $<0.05 \mu g/L$. Similar concentrations have been reported for these two PAHs [18]. Masoner et al. [20] studied leachates with very 358 different waste compositions, ages and geographic/climatic characteristics, and 359 determined higher concentrations for both PAHs with median values of 5 µg/L for 360 naphthalene and 1 µg/L for phenanthrene. DEET (N,N-dimethyl-m-toluamide) is a 361 362 relevant compound frequently reported in leachates at high levels, e.g. 11-202 µg/L 363 [13], 0.03-144 µg/L [19] or 70 µg/L (median value) [20]. The concentration found for 364 DEET in raw leachate studied in this work was low $(1.61 \text{ }\mu\text{g/L})$ in comparison to these 365 previous studies. In the case of triphenyl phosphate, the concentration reported by 366 Masoner et al. [20] is quite similar (2 μ g/L). Tonalide has been pointed out as one of the 367 most abundant musk compounds in the environment [1] but it was not detected before 368 in former works about leachates.

369

It must be noticed that despite the analytical characterization included the monitoring

370 of 115 target compounds and around 1300+900 suspect compounds, only a small

371 percentage of the compounds included in the study were finally detected. This can be

372 explained, firstly, because the applied analytical methods were quite generic in order to

373 <u>cover a wide range of OMCs. Thus, it can be expected that these conditions do not suits</u>

in the same way for all the analytes, but it is an adequate approach to carry out such a
comprehensive analysis. On the other hand, the OMC characterization was limited to
urban landfill leachates and OMC content can be considerably different depending on
the type of landfill, the age, etc.

378 With respect to the effectiveness of the treatment line to remove the confirmed compounds, a summary of the results is shown in Fig. 4 and $\frac{5A.4}{A}$. The full scan 379 380 chromatograms obtained for the coagulated-flocculated and SPF effluent (Fig. 5A.4) indicates a clear reduction of the area of the peaks. This can be interpreted as a general 381 382 reduction of the components of the different effluents, including organic matter, OMCs 383 and other unknown compounds that could be present. In particular, for the suspect 384 compounds confirmed with standards, a reduction of the concentration was observed 385 after the SPF treatment (89%), with an elimination of 92% considering the complete 386 treatment line (Fig. 4). The remaining concentration of di-n-nonyl phthalate was somehow high in the SPF effluent (1.18 μ g/L) (Table A.3), despite a substantial 387 388 elimination of 81% compared with the initial load.

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390 4. Conclusions

391 The detailed analysis of the leachates obtained after each stage of the treatment line 392 confirmed that a clear reduction of the OMC content was obtained. The results of the target analysis by LC-QqLIT-MS/MS are the most relevant due to the number of 393 394 monitored and quantified compounds (115) and indicated a reduction of the 94% of the 395 initial OMC load (from the raw to the IBR effluent). The additional suspect analysis by 396 LC-QTOF-MS and GC-Q-MS permitted to evaluate 16 compounds not determined by LC-QqLIT-MS/MS, demonstrating that they were also eliminated in the range from 397 398 21% to >99%. LC-QTOF-MS and GC-Q-MS also revealed the presence of nonpreviously reported OMCs in leachates (di-n-nonyl phthalate, o-phenylphenol, and 399 400 tonalide). It has been demonstrated that raw leachate treated by coagulation/flocculation 401 and SPF can be successfully incorporated to a conventional aerobic biological treatment 402 after the specific adaptation of the activated sludge. The overall elimination of the quantified OMCs was 94%. Therefore, leachate treatment based on the combination of 403 404 different steps (solid separation, advanced oxidation and adapted biotreatment) could be 405 successful not only in COD removal and toxicity reduction, but also in a substantial

406 OMC elimination that deserves to be demonstrated by a proper advanced analytical407 technique train.

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416 Appendix A. Supplementary data

417 Supplementary material related to this article can be found in the online version.

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Fig. 1. DOC elimination throughout the solar photo-Fenton and continuous aerobic biological treatments. H_2O_2 consumption during SPF and TN decrease along the biotreatment are also shown.

- 569 Fig. 2. Summary of the results obtained in the target analysis by LC-QqLIT-MS/MS:
- total OMC load (μ g/L), percentage of removal and number of detected OMCs.
- 571 Fig. 3. Comparison of the total ion chromatograms (TIC) obtained in the full scan mode
- 572 by LC-QTOF-MS of raw and treated leachate (coagulated-flocculated, photo-Fenton
- and aerobic immobilized biomass reactor (IBR) effluent).
- Fig. 4. Evolution of the concentrations of the OMCs confirmed by the suspect analysisusing GC-Q-MS.
- 576 **Fig. 5.** Comparison of the total ion chromatograms (TIC) obtained in the full scan mode
- 577 by GC-Q-MS of coagulated-flocculated and solar photo-Fenton effluent (total area is
- 578 also shown).

Table 1Summary of the results obtained in the suspect analysis by LC-QTOF-MS.

Leachate	Compound	Formula [M]	RT ^a (min)	$\begin{bmatrix} \mathbf{M} + \mathbf{H} \end{bmatrix}^+$ (m/z)	Mass error (ppm)	IRD ^b (%)	MS/MS library score (%)	>2 PIs ^c (<5 ppm)	Standard purchased	Confirmed by standard
Coagulated-flocculated effluent	Chlorfenvinphos	$C_{12}H_{14}Cl_3O_4P$	10.37	358.9768	-2.6	1.7	Sciex (97) Massbank (99)	Yes	Yes	Yes
	Niflumic acid	$C_{13}H_9F_3N_2O_2$	9.77	283.0688	-0.8	3.1	Sciex (93) Massbank (89)	Yes	Yes	Yes
	Tramadol	$C_{16}H_{25}NO_2$	5.94	264.1958	-1.1	2.7	Sciex (90) Massbank (99)	Yes	Yes	Yes
	Valsartan	$C_{24}H_{29}N_5O_3$	8.95	436.2343	-0.2	0.6	Sciex (91) Massbank (99)	Yes	Yes	Yes
IBR ^d effluent	Tramadol	C ₁₆ H ₂₅ NO ₂	5.93	264.1958	-0.1	0.6	Sciex (99) Massbank (99)	No	Yes	Yes
	Valsartan	$C_{24}H_{29}N_5O_3$	8.95	436.2343	0.7	3.0	Massbank (73) Chemspider (90)	Yes	Yes	Yes

^aRT: retention time; ^bIRD: isotopic ratio difference; ^cPI: product ion; ^dIBR: Aerobic immobilized biomass reactor.

Table 2

Summary of the results obtained in the suspect analysis by GC-Q-MS (data shown for the coagulated-flocculated effluent).

Compound	Stage of the leachate treatment	Type of compound	Previously reported in leachates	Formula [M]	RT ^a (min)	MS library match (%)	Standard purchased	Confirmed by standard
Bisphenol A	C/F ^b	Plastizier	YES	$C_{15}H_{16}O_2$	23.92	89	YES	NO
Butyl-benzyl phthalate	C/F	Plastizier (EPA ^c Priority list)	YES	$C_{19}H_{20}O_4$	27.15	88	YES	YES
Chrysene	C/F	PAH ^d (POPs ^e , EPA Priority list)	YES	$C_{18}H_{12}$	28.44	<50	YES	NO
2,4-Dichlorophenol	C/F	Intermediate (EPA Priority list)	YES	$C_6H_4Cl_2O$	5.23	80	YES	YES
$DEET^{f}$	Raw, C/F, SPF ^g , IBR ^h	Insect repellent	YES	$C_{12}H_{17}NO$	9.82	97	YES	YES
Diisobutyl phthalate	C/F	Plastizier	YES	$C_{16}H_{22}O_4$	15.95	91	YES	YES
Di-n-butyl phthalate	C/F	Plastizier (EPA Priority list)	YES	$C_{16}H_{22}O_4$	18.43	90	YES	YES
Di-n-nonyl phthalate	Raw, C/F, SPF, IBR	Plastizier	NO	$C_{26}H_{42}O_4$	33.92	83	YES	YES
Diphenylamine	C/F	Pesticide	YES	$C_{12}H_{11}N$	10.51	90	YES	YES
Fluorene	C/F	PAH (POPs, EPA Priority list)	YES	$C_{13}H_{10}$	9.93	79	YES	NO
Hexachlorobenzene	C/F	Pesticide (POPs, EU ⁱ Priority list)	YES	C_6Cl_6	12.38	<50	YES	NO
Naphthalene	C/F	PAH (POPs, EPA Priority list, EU Priority list)	YES	$C_{10}H_{8}$	5.39	65	YES	YES
Naphthalic anhydride	C/F	Pesticide	NO	$C_{12}H_6O_3$	20.13	73	YES	NO
Phenanthrene	Raw, C/F	PAH (POPs, EPA Priority list)	YES	$C_{14}H_{10}$	13.82	82	YES	YES
o-Phenylphenol	C/F	Pesticide	NO	$C_{12}H_{10}O$	8.80	87	YES	YES
Pyrimethanil	C/F	Pesticide	NO	$C_{12}H_{13}N_3$	14.14	72	YES	NO
Sulfur (S8)	C/F	Pesticide	YES	-	20.05	55	YES	NO
Tonalide	Raw, C/F, SPF, IBR	Musk compound	NO	$C_{18}H_{26}O$	10.96	80	YES	YES
Triphenyl phosphate	C/F	Flame retardant, plastizier	YES	$C_{18}H_{15}O_4P$	27.71	83	YES	YES

^aRT: retention time; ^bC/F: coagulated-flocculated effluent; ^cEPA: Environmental Protection Agency; ^dPAH: polycyclic aromatic hydrocarbon; ^ePOP: persistent organic pollutant; ^fDEET: *N*,*N*-Dimethyl-m-toluamide; ^gSPF: effluent from solar photo-Fenton; ^hIBR: effluent from aerobic immobilized biomass reactor; ⁱEU: European Union.









Highlights

- Addition of landfill leachates treated by solar photo-Fenton to biological systems.
- Treatment effectiveness: evaluation of organic microcontaminant (OMC) removal.
- Evaluation using target and suspect analysis by LC-MS/MS, LC-QTOF-MS and GC-MS.
- Treated leachate: reduction of 94% of the initial quantified OMC concentration.
- Detection of 3 non-reported OMCs: di-*n*-nonyl phthalate, *o*-phenylphenol and tonalide.

STATEMENT OF NOVELTY

This study describes the development of a specific treatment for landfill leachates using a solar photo-Fenton process to reduce toxicity and increase biodegradability, and a subsequent biological treatment. The treatment effectiveness was evaluated considering organic microcontaminant removal, which is highly novel regarding the few related works. This information is also relevant for the potential design of specific treatments for leachates. A combined target and suspect strategy were applied using low and high-resolution mass spectrometry tools. The treated leachate shows a reduction of 94% of the initial quantified OMC concentration. 3 non-reported OMCs were detected: di-*n*-nonyl phthalate, *o*-phenylphenol and tonalide.

Revised Supplementary Material Click here to download Supplementary Material: Revised Supplementary_Material.pdf



1 Abstract

Conventional wastewater treatments are not usually effective in the remediation of 2 specific landfill leachates due to their high content in toxic and recalcitrant compounds. 3 Advanced and intensive treatments are needed for the decontamination and possible 4 recycling of these effluents. In this studyHere, the combination of advanced oxidation 5 processes (solar photo-Fenton) and an aerobic biological reactor have been applied to 6 7 treat urban landfill leachates. The effectiveness of the proposed treatment line was also evaluated considering the removal of organic microcontaminants (OMCs) identified in 8 9 the different phases, which is an innovative practice. The analytical strategy included: 10 (i) a target approach (115 analytes) by liquid chromatography-mass spectrometry (LC-11 MS/MS); and two suspect approaches using (ii) LC-high-resolution MS (database with >1300 compounds) and (iii) gas-chromatography-MS (database with >900 compounds). 12 13 OMCs on the treated landfill leachate was reduced up to 94% of the initial quantified concentration. 8Eight target compounds (mainly pharmaceuticals) out of 155 target 14 15 analytes represented 85% of theis OMC concentration in the raw leachate: cotinine, diclofenac, gabapentin, ketoprofen, lidocaine, mecoprop, nicotine and trigonelline. 16 17 Three <u>3</u> non-previously reported OMCs were confirmed: di-n-nonyl phthalate, ophenylphenol and tonalide. The study demonstrated that lLeachate partially oxidized by 18 solar photo-Fenton process can be successfully incorporated to biological systems to 19 complete the treatment by means of specifically adapted biomass. 20

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