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1 **Nitrogen Metabolism and Yield Response to Increases in Nitrogen-Phosphorus**
2 **Fertilization: Improvement in Greenhouse Cultivation of Eggplant (*Solanum***
3 ***melongena* Cv. Bonica)**
4

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8

9 **ABSTRACT**

10 The effect that application of nitrogen-phosphorus (NP) rates exerts on some parameters of nitrogen
11 metabolism and on yield in aubergine plants (eggplant, *Solanum melongena* cv. Bonica) was studied. All
12 plants were grown under controlled conditions in an experimental greenhouse. The treatments consisted
13 of the combination of three rates of N in the form of KNO₃ (N₁: 15 g m⁻², N₂: 22.5 g m⁻² and N₃: 30 g m⁻²)
14 together with two rates of P in the form of H₃PO₄ (P₁: 24 g m⁻² and P₂: 36 g m⁻²), for a total of six
15 treatments. Results indicate that regardless of the P rate applied, an increase in the N fertilizer increased
16 NO₃⁻ assimilation as well as total and commercial yield. Use of the P₂ rate favored all of these processes
17 in comparison with the results obtained with the P₁ rate, in addition to notably reducing the foliar levels
18 of NO₃⁻ and noncommercial yield.
19

20 **Keywords:** *Solanum melongena* L.; NP fertilization; nitrogen assimilation; yield
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22

23 **INTRODUCTION**

24 Plant growth is dependent on an adequate nitrogen (N) supply in order to form amino acids,
25 proteins, nucleic acids, and other cellular constituents. For most plants, inorganic N is obtained from
26 the soil in the form of nitrate (NO₃⁻-N). It has long been recognized that the rate-limiting step for N
27 assimilation, the reduction of NO₃⁻-N to nitrite (NO₂⁻-N) catalyzed by nitrate reductase (NR), is
28 highly regulated (Solomonson and Barber, 1990; Huber et al., 1996; Sivasankar and Oaks, 1996).
29 Nitrate availability, growth regulators, light, products of NO₃⁻-N assimilation and other
30 physiological and environmental parameters are all factors in regulation of NO₃⁻-N assimilation
31 (Campbell, 1988; Lillo, 1994; Kaiser and Huber, 1994; Crawford, 1995; Sivasankar and Oaks,
32 1996).

33 Appropriate levels of NO₃⁻-N derived from proper N fertilization increases the amount and activity
34 of NR; this leads to a corresponding increase in the potential for NO₃⁻-N reduction and confers a
35 greater capacity for general amino-acid synthesis, protein synthesis, or total N assimilation
36 (Vincentz et al., 1993; Migge and Becker, 1996; Barneix and Causin, 1996). In addition, it has been
37 shown that plant growth and yield rates are often dependent on N supply (Mattson et al., 1991).
38 Nevertheless, excessive N fertilization can be harmful, and NO₃⁻-N accumulation is a problem in
39 most crops (Steingröver et al., 1982; Blom-Zandstra and Lampe, 1985; Van der Boon et al., 1990;
40 Quillere et al., 1994). Leaf tissues accumulate NO₃⁻-N when the rate of NO₃⁻-N delivery exceeds the
41 NO₃⁻-N assimilation capacity (Mackown et al., 1990; Quillere et al., 1994). In addition, nitrogen
42 fertilizers contribute to groundwater and surface-water pollution through leaching and soil erosion
43 (Sisson et al., 1991).

44 The role of P in N metabolism has been studied in detail, although almost always from the
45 standpoint of the effect of deficiencies in this element. Nitrate assimilation is altered when plants

46 are deprived of P. The most commonly described effects are: (i) reduction in NO_3^- -N absorption
47 through the roots (Rufty et al., 1990; Rufty et al., 1991; Pilbeam et al., 1993), possibly due to
48 decreased availability of radicular ATP and a limitation in synthesis of the membrane transport
49 system for NO_3^- -N (Rufty et al., 1993); (ii) decrease in the translocation of NO_3^- -N from the roots
50 to the aerial part (Rufty et al., 1990; Pilbeam et al., 1993), related to a fall in water pressure via the
51 root and xylem (Rufty et al., 1993); and (iii) increases in amino acid levels (Rufty et al., 1990; Rufty
52 et al., 1993). These factors lead to a decreased activity of NR and therefore, in the assimilation of
53 NO_3^- -N (Pilbeam et al., 1993).

54 In the present work, we analyzed the effect of N(NO_3^- -N) and P (H_3PO_4) fertilization on some
55 parameters of N metabolism and on yield of eggplant, the latter being the final indicator of plant
56 growth. Our aim was to determine the most appropriate rates of these elements for eggplant grown
57 under greenhouse conditions.

58

59 MATERIALS AND METHODS

60 **Crop Design.** *Solanum melongena* cv. Bonica were seeded in cell flats (cell size $3 \times 3 \times 10$ cm)
61 filled with peat-lite mixture, placed on benches under the greenhouse conditions described below,
62 for a period of 8 weeks; then seedlings were transplanted and grown under controlled conditions in
63 an experimental greenhouse at Centro de Investigación y Desarrollo Hortícola, El Ejido, Almería,
64 Spain. The experiment was conducted from 1993 to 1995. The climate is semiarid and the lands are
65 intensively used for agriculture. The soil used was loamy-sand with the following characteristics:
66 sand (37.3%), silt (48.6%), and clay (10.1%), CaCO_3 equivalent (26.82%), CaCO_3 active (14.35%),
67 total N (3.5 g kg^{-1}), total organic C (36.1 g kg^{-1}), PO_4^{3-} (890 mg kg^{-1}), K^+ (5.34 g kg^{-1}), pH (H_2O ,
68 8.45; KCl; 8.01), electrical conductivity (EC) 4.63 dS m^{-1} . The relative humidity was 60-80% and
69 the temperature range 24 ($4 \text{ }^\circ\text{C}$ with extremes 15 and $30 \text{ }^\circ\text{C}$ in the greenhouse. The experimental
70 design was a factorial arrangement in a randomized complete block with six treatments.

71 Container-grown eggplants were transplanted into two rows 100 cm apart and trickle irrigated. Each
72 treatment was replicated four times in four individual plots of $4 \text{ m} \times 2 \text{ m}$ wide (24 plots). Each plot
73 contained eight treated plants. The irrigation water had the following properties: pH, 8.05; EC 2.03
74 dS m^{-1} ; Cl^- 483.90 mg L^{-1} ; Na^+ 305.76 mg L^{-1} ; K^+ 10.16 mg L^{-1} ; HCO_3^- 278.15 mg L^{-1} .

75 The different treatments consisted of applying increasing rates of both N and P in the following
76 manner: N in the form of KNO_3 (N_1 : 15 g m^{-2} , N_2 : 22.5 g m^{-2} , and N_3 : 30 g m^{-2}) and P in the form
77 of H_3PO_4 (P_1 : 24 g m^{-2} and P_2 : 36 g m^{-2}). Calcium (11 g m^{-2}) and magnesium (3 g m^{-2}) were supplied
78 as sulfates. Each nutrient was applied gradually at the specified rate with water irrigation over the
79 entire growth period of the plants. Fertilization-irrigation was complemented with the following
80 micronutrients: Fe: 0.5 mg L^{-1} ; B: 0.1 mg L^{-1} ; Mn: 0.1 mg L^{-1} ; Zn: 0.075 mg L^{-1} ; Cu: 0.075 mg L^{-1}
81 and Mo: 0.05 mg L^{-1} . The pH values of the solution oscillated between 5 and 6; Fe was applied as
82 FeEDDHA, B as H_3BO_3 , and the remaining micronutrients as sulfates.

83 **Plant Sampling.** Leaf samples were taken only from plants with fully expanded leaves of the
84 same size ($> 25 \text{ cm}$). Leaves were picked about one-third of the plant height from the plant apex
85 during flowering (95 days after transplanting until 170 days) every 2 weeks (Lo'pez-Cantarero et
86 al., 1995). Leaves were rinsed three times in distilled water after being disinfected with nonionic
87 detergent at 1% (Wolf, 1982), then blotted on filter paper. At each sampling, fresh leaf matter was
88 used for the NRA assay, amino acids and proteins; then a subsample was dried in a forced air oven
89 at $70 \text{ }^\circ\text{C}$ for 24 h, ground in a wiley mill, and then placed in plastic bags for further analyses (NO_3^-
90 and organic N).

91 **Plant Analysis.** *NO_3^- Determination.* NO_3^- -N was analyzed from an aqueous extraction of 0.2 g of
92 dried and ground leaf material in 10 mL of Millipore-filtered water; $100 \text{ } \mu\text{L}$ aliquot was taken for
93 NO_3^- -N determination and added to 10% (w/v) salicylic acid in sulfuric acid at 96%, and the NO_3^- -

94 N concentration was measured by spectrophotometry as performed by Cataldo et al. (1975). The
95 results were expressed as mg g^{-1} dry weight (dw).

96 *Detection of in Vivo NR Activity.* The basic method was an adaptation of the in vivo NR assay by
97 Jaworski (1971) and Maurin^o et al. (1986). Leaves were cut into 5 mm sections and the sample (0.5
98 g) was placed in 10 mL incubation buffer (100 mM potassium phosphate buffer, pH 7.5) and 1%
99 (v/v) propanol. The sample was infiltrated and the intracellular spaces of the tissues were flushed
100 with buffer using a vacuum (0.8 bar). After 10 min, the vacuum was released and the samples were
101 reevacuated. The samples were incubated at 30 °C in darkness for 1 h and placed in a boiling water
102 bath to stop the NR activity. The resulting NO_2^- was measured by the method of Snell and Snell
103 (1949), and the NR activity was expressed as $\mu\text{mol NO}_2^- \text{g}^{-1}$ fresh weight (fw) h^{-1} .

104 *Amino Acids and Soluble Proteins Determination.* Fresh leaf samples (0.5 g) were crushed with
105 cold phosphate buffer (50 mM KH_2PO_4 , pH 7) and centrifuged at 12 000g for 15 min. The resulting
106 supernatant was used for the determination of total amino acids by the ninhydrin method as
107 described by Yemm and Cocking (1955); total free amino acids were expressed as mg glycine g^{-1}
108 fw. Soluble proteins were measured by Bradford G-250 reagent (Bradford, 1976) and expressed as
109 $\text{mg bovine serum albumin g}^{-1}$ fw.

110 *Organic N Determination.* A 0.1 g dw subsample was digested with sulfuric acid with H_2O_2 (Wolf,
111 1982). After dilution with deionized water, a 1 mL aliquot of the digest was added to the reaction
112 medium containing buffer (5% potassium sodium tartrate, 100 μM sodium phosphate, and 5.4%
113 [w/v] sodium hydroxide), 15/0.03% (w/v) sodium salicylate/sodium nitroprusside, and 5.35% (v/v)
114 sodium hypochlorite. Samples were incubated at 37 °C for 15 min and organic N was measured by
115 spectrophotometry according to Baethgen and Alley (1989). The results were expressed as mg g^{-1}
116 dw.

117 *Yield.* Plant yield was expressed as the mean of fruit weight. Collected eggplants were weighed on
118 each plant at harvest. Commercial yield represents fruits with acceptable color and caliber, while
119 those without acceptable features represent noncommercial or residual yield. Total yield (kg plant^{-1})
120 is the sum of both types of yield.

121 **Statistical Analysis.** The data were submitted to three statistical treatments in order to determine:
122 (i) differences between N rates at each P level; (ii) differences between P rates at each N level; and
123 (iii) differences between all fertilizer treatments. Standard analysis of variance (ANOVA) techniques
124 were used to assess the treatment means. Where there were no climatic effects on yield, so ANOVAs were
125 performed on pooled data from both years. Treatment means were compared using the least
126 significant difference at the 0.05 probability level. The mean of separation according to the Duncan's
127 Multiple Range test is represented with letters in the tables and figures. When appropriate, a simple
128 regression analysis was performed between parameters: $p < 0.05$) *, $p < 0.01$) **, ns) not
129 significant.

130 RESULTS AND DISCUSSION

131 **Response to the N Rate.** When the P rate was P_1 , most of the N-metabolism parameters studied, and
132 yield, were influenced directly by the N rate applied. One of the major and limiting stages of NO_3^-
133 assimilation is NR activity (Huber et al., 1996; Sivasankar and Oaks, 1996). Figure 1 presents the values
134 obtained in the *in vivo* assay of NR activity, presenting the maximal activity when the N_3 rate was used
135 with a 12% increase with respect to the minimal activity found with N_1 . On the other hand, the NO_3^-
136 levels (Table 1) showed a behavior contrary to the NR activity (Figure 1), given that the maximal foliar
137 levels were reached with N_1 (8.43 mg g^{-1} dw), while the minimum resulted from N_3 (7.29 mg g^{-1} dw).
138 One of the principal factors regulating both the increase in *de novo* NR synthesis and its activity is the
139 presence of NO_3^- (Campbell, 1988; Kaiser and Huber, 1994; Crawford, 1995; Sivasankar and Oaks,
140 1996). In view of this fact, an increase in N fertilization (as caused by the treatments N_2P_1 and N_3P_1)
141 would raise the foliar levels of NO_3^- . This should stimulate NR activity (Figure 1) and explain the

142 decreased foliar NO_3^- content in these treatments (Table 1) due to its reduction. The relationship between
143 the two parameters in our experiment proved negative (r^2 , 0.754**). With respect to the nitrogenous
144 compounds of high and low molecular weight, such as amino acids and proteins, these are the principal
145 products of NO_3^- assimilation (Barneix and Causin, 1996). When the P rate was P_1 , we found no effect of
146 the different N rates on the foliar amino acid levels (Table 1). On the contrary, the protein levels (Table
147 1) reflected the greatest reduction of NO_3^- , since the N_2P_1 and N_3P_1 treatments registered the maximal
148 concentrations. Organic nitrogen was another parameter that strengthened with intensified NO_3^-
149 reduction (Vincentz et al., 1993). Compared to the amino acids and proteins, the organic-nitrogen
150 concentration (Table 1) augmented proportionally to the increase in NR activity (Figure 1), given that
151 the maximal concentration was found when the nitrogen rate was N_3 (47.84 mg g^{-1} dw). The relationship
152 between the NR activity and the organic nitrogen concentration was positive (r^2 0.827**), and there was
153 a close relationship between the NO_3^- reduction and the organic-nitrogen content.

154 Nitrogen fertilization plays an important role in the growth and development of most plants (Silberbush
155 and Lips, 1991; Mattson et al., 1991). Yield (g plant^{-1}) was clearly influenced by the nitrogen rate applied,
156 since the maximum values in total as well as commercial and noncommercial yield were recorded for
157 the N_3 treatment (Table 2). Increased N fertilization raised both the noncommercial and commercial
158 yields (Table 2).

159 When the P rate was P_2 , the trend described above varied for some parameters, indicating a
160 possible influence of P. The NR activity (Figure 1) was again influenced by N fertilization, since as
161 the N rate increased, so did enzymatic activities; N_3 registered a 16% increase with respect to the
162 minimal activity values of N_1 . In relation to NO_3^- (Table 1), in contrast to the previous case (P_1), no
163 decrease was recorded in the foliar level in proportion to increased NR activity, given that N_1P_2 and
164 N_3P_2 gave the minimal level (6.86 mg g^{-1} dw). In this case, these parameters showed no relationship
165 ($r^2=0.271$ ns), indicating that a higher P application could influence foliar NO_3^- levels.

166 Amino acids and proteins (Table 1) showed a trend similar to that observed when the P fertilization
167 was P_1 , given that no clear effect was found of the N rate on these levels, as differences between
168 treatments were minimal. Finally, organic N (Table 1), as was true when the P rate was P_1 , proved
169 strongly influenced by N, the maximal concentration appearing at N_3 (48.82 mg g^{-1} dw) and the
170 minimum at N_1 (45.39 mg g^{-1} dw). The relationship between the NR activity and the organic-nitrogen
171 concentration was again positive ($r^2=0.799$ **), indicating the lack of interference of P in NO_3^-
172 assimilation

173 Yields were directly dependent on the N rate applied (Table 2), given that N_3 gave the maximum
174 values (18.67 and 17.57 kg plant^{-1} , respectively). It is striking that noncommercial yield differed
175 from the rest in that the minimal value was recorded for the N_3 rate (1.10 kg plant^{-1}), showing a trend
176 completely opposite that shown when the P rate was P_1 , and thus, increased P fertilization appears
177 to diminish noncommercial yield.

178 In this sense, we conclude that regardless of the rate of P applied, the N fertilization exerts direct
179 influence on NO_3^- assimilation with increases in NR activity, on the foliar concentration of organic
180 N (Figure 1; Table 1), and on both total and commercial yield (Table 2). The data suggest that
181 increased fertilization did not result in N intoxication as a result of (i) the low foliar NO_3^- levels
182 found at N_3 , indicating that levels of this ion that would saturate or exceed assimilation capacity
183 were not reached (Mackown et al., 1990); and (ii) the low amino-acid levels (Table 1), indicating
184 that N concentrations were not great enough to raise amino acid levels and thereby inhibit NO_3^-
185 reduction (Barneix and Causin, 1996; Padgett and Leonard, 1996; Sivasankar and Oaks, 1996).
186 Finally, P_2 played a part in the foliar NO_3^- content and noncommercial yield, decreasing the levels
187 of both (Tables 1 and 2).

188
189 **Response to P Rate.** Regardless of the N level applied, the use of P_2 increased the NR activity
190 (Figure 2) in comparison with the activity shown by this enzyme when the P rate was P_1 , this increase
191 in all cases exceeded 10%. The foliar NO_3^- content (Table 3) also showed an influence by the P_2

192 fertilization, since regardless of the N level used, the application of P₂ decreased NO₃⁻ levels. It is
193 possible that the larger P application reversed the processes described in the Introduction in relation
194 to P deprivation as well as the effect of this element on N metabolism, thereby increasing absorption,
195 translocation toward the aerial part, and NO₃⁻ assimilation (Rufty et al., 1993; Pilbeam et al., 1993).
196 This would explain the increased NR activity (Figure 2) and the decrease in the NO₃⁻ level (Table
197 3).

199 With respect to the amino acids and proteins (Table 3), the P rate had no appreciable influence,
200 since either there was no statistical significance in the case of the amino acids or the behavior was
201 variable in the case of the proteins. On the contrary, the organic-N concentration (Table 3) increased
202 with P₂, due possibly to the strengthening effect of the P rate on NO₃⁻ assimilation. Finally, in relation
203 to yield, P₂ showed a negative effect on total and commercial yield when the N rate was N₁,
204 decreasing the values of both (Table 4), while for the N₂ and N₃ rates, an increase in P raised total
205 and commercial yield (Table 4). The P₂ rate caused noncommercial yield to fall sharply and
206 significantly regardless of the N level (Table 4).

207 **Response to NP Fertilization.** This aspect of our study was undertaken to compare statistically
208 all of the fertilization treatments, in order to identify the most adequate combination or
209 combinations. The values of NR activity (Figure 3) show that the minimum activity was given by
210 N₁P₁, indicating that the combination of low rates of N (N₁ and N₂) and P (P₁) did not increase NR
211 activity. On the contrary, the application of N₃P₂ produced greater NR activity (Figure 3), with an
212 increase of 28% over the minimum. With respect to the NO₃⁻ values (Table 5), the treatments N₁P₂,
213 N₂P₂, and N₃P₂ gave the lowest levels, with maximal decreases of 19% with respect to the maximum
214 presented in the treatment N₁P₁.

215 Amino acids and proteins are the main products of NO₃⁻ assimilation (Barneix and Causin, 1996),
216 but these did not show a clear effect of the treatments, and thus are not useful parameters for
217 selecting the most adequate fertilization treatments. On the other hand, the organic N did present a
218 positive relationship with NO₃⁻ reduction, since the maximal concentration was presented in the
219 combination N₃P₂ and the minimum in N₁P₁ (Table 5).

220 Finally, the total and commercial yield (Table 6) was maximal in N₃P₂, with increases of 13 and
221 14%, respectively in comparison with the minimal yield obtained in the N₁P₂ treatment. Meanwhile,
222 noncommercial yield registered its minimal value in N₃P₂, with a decline of 44% from the maximal
223 value of treatment N₃P₁.

224 Taking into account the results of these experiments, we conclude that the most adequate fertilizer
225 combination for eggplant was N₃P₂. This treatment gave the greatest NO₃⁻ assimilation (minimal
226 NO₃⁻ levels; maximal NR activity, and maximal organic-N concentration), largest total and
227 commercial yield and least noncommercial yield. Finally, P₂ is noteworthy for strengthening all of
228 the parameters and processes mentioned above, and thus, it appears that an increased P application
229 before high N rates can reduce noncommercial in favor of commercial yield, and can avoid or reduce
230 both NO₃⁻ pollution as well as the negative effects of NO₃ accumulation in different plant organs.
231

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Table 1. Accumulation of Various N Forms due to Application of Different N Rates^a

treatments		mg g ⁻¹ fw		mg g ⁻¹ dw	
		amino acids	proteins	organic N	nitrate
P ₁	N ₁	1.37 a	25.8 b	41.1 c	8.43 a
	N ₂	1.37 a	27.4 a	43.4 b	7.80 b
	N ₃	1.38 a	27.3 a	47.8 a	7.29 c
P ₂	N ₁	1.36 a	26.3 b	45.4 c	6.86 a
	N ₂	1.35 a	27.4 a	46.6 b	7.01 a
	N ₃	1.34 a	26.1 b	48.9 a	6.86 a

^aValues followed by the same letter within a column and phosphorus treatment were not measurably different (p<0.05).

Table 2. Effect of N Rate on Yield of Eggplant^a

treatments		yield (kg plant ⁻¹)		
		total	commercial	noncommercial
P ₁	N ₁	17.7 b	15.9 c	1.83 b
	N ₂	17.9 b	16.1 b	1.86 b
	N ₃	18.3 a	16.4 a	1.97 a
P ₂	N ₁	16.3 c	15.2 c	1.16 a
	N ₂	17.9 b	16.7 b	1.16 a
	N ₃	18.7 a	17.6 a	1.10 b

^aValues followed by the same letter within a column and phosphorus treatment were not measurably different (p<0.05).

Table 3. Accumulation of N Compounds due to Application of the Different P Rates^a

treatments		mg g ⁻¹ fw		mg g ⁻¹ dw	
		amino acids	proteins	organic N	nitrate
N ₁	P ₁	1.37 a	25.8 b	41.1 b	8.43 a
	P ₂	1.36 a	26.3 a	45.4 a	6.86 b
N ₂	P ₁	1.37 a	27.4 a	43.4 b	7.80 a
	P ₂	1.35 a	27.4 a	46.6 a	7.01 b
N ₃	P ₁	1.38 a	27.3 a	47.8 b	7.29 a
	P ₂	1.34 a	26.1 b	48.9 a	6.86 b

^aValues followed by the same letter within a column and nitrogen treatment were not measurably different (p<0.05).

Table 4. Effect of P Rate on Yield of Eggplant^a

treatments		yield (kg plant ⁻¹)		
		total	commercial	noncommercial
N ₁	P ₁	17.7 b	15.9 a	1.83 a
	P ₂	16.3 b	15.2 b	1.16 b
N ₂	P ₁	17.9 a	16.1 b	1.86 a
	P ₂	17.9 a	16.7 a	1.16 b
N ₃	P ₁	18.3 a	16.4 b	1.97 a
	P ₂	18.7 a	17.6 a	1.10 b

^aValues followed by the same letter within a column and nitrogen treatment were not measurably different (p<0.05).

Table 5. Accumulation of N Compounds due to Application of the Different NP Rates^a

treatments	mg g ⁻¹ fw		mg g ⁻¹ dw	
	amino acids	proteins	organic N	nitrate
N ₁ P ₁	1.37 a	25.8 c	41.1 e	8.43 a
N ₂ P ₁	1.37 a	27.4 a	43.4 d	7.80 b
N ₃ P ₁	1.38 a	27.3 ab	47.8 b	7.29 bc
N ₁ P ₂	1.36 a	26.3 b	45.4 cd	6.86 c
N ₂ P ₂	1.35 a	27.4 a	46.6 c	7.01 c
N ₃ P ₂	1.34 a	26.1 b	48.9 a	6.86 c

^aValues followed by the same letter within a column were not measurably different (p<0.05).

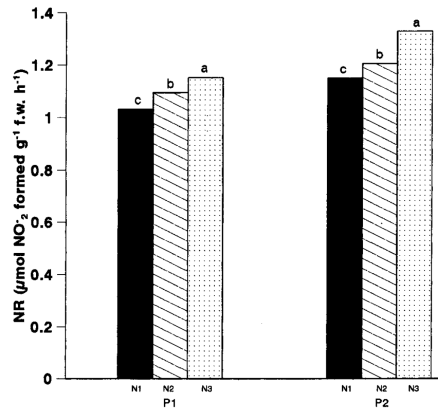


Figure 1. Effect of the different N rates (N₁ solid bar, N₂ striped bar, N₃ dotted bar) on the in vivo nitrate-reductase activity. For each P rate, different letters indicate a significant effect (p < 0.05) according to Duncan's test

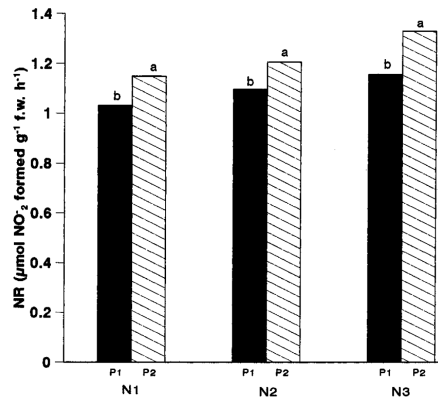


Figure 2. Effect of the different P rates (P₁ solid bar, P₂ striped bar) on the in vivo nitrate-reductase activity. For each N rate, different letters indicate a significant effect (p < 0.05) according to Duncan's test.

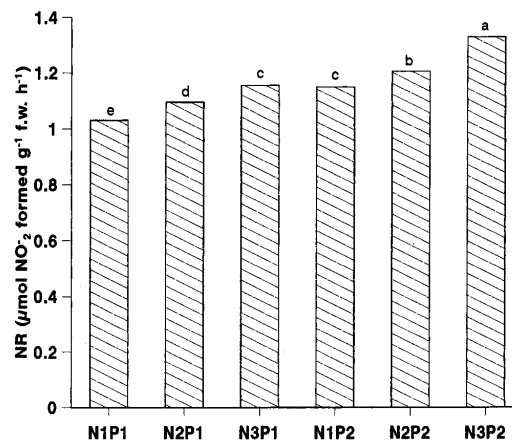


Figure 3. Effect of NP rates on the in vivo nitrate-reductase activity. Different letters indicate a significant effect (p < 0.05) according to Duncan's test.